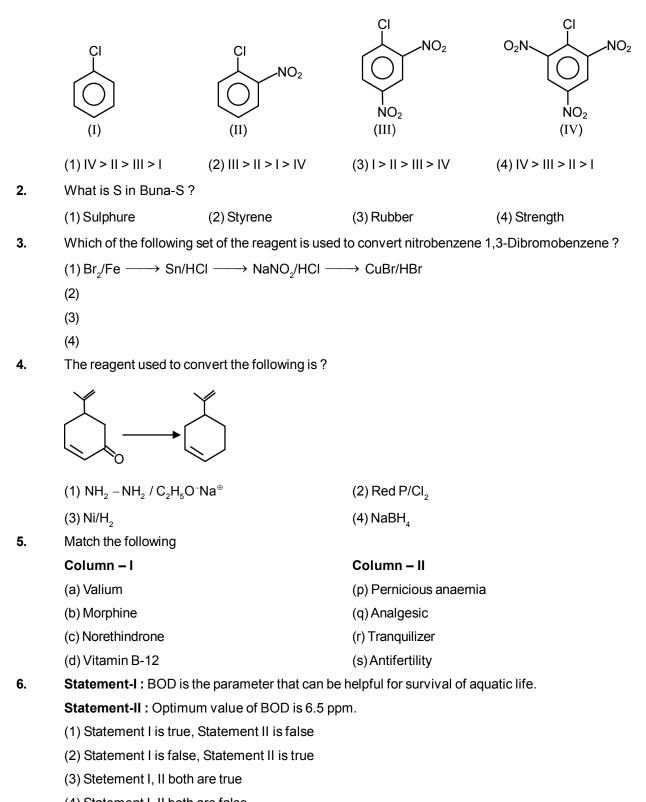
## MOMENTUM

CHEMISTRY

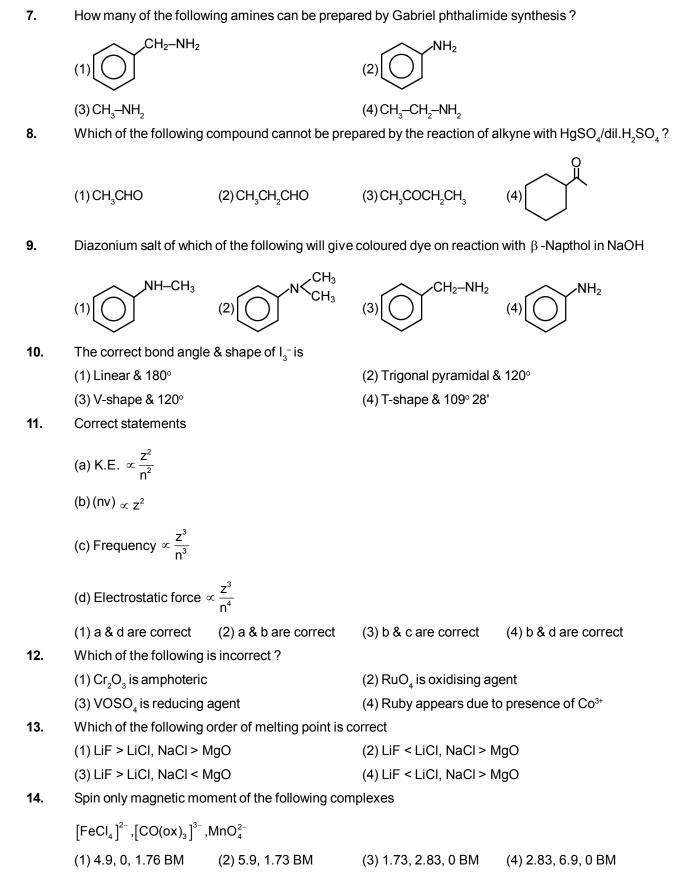
(24 Feb 2021) Shift-2

1. Compare the rate of aromatic nucleophilic substitution reaction of the following compounds



(4) Statement I, II both are false

## MOMENTUM



## MOMENTUM

<b>15.</b> $\alpha$ – sulphur, $\beta$ – sulphur, $S_2 \rightarrow$ find how many are paramag	netic
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16. Which of the following can be used for clotting of blood efficiently?
(1) NaHCO<sub>3</sub>
(2) FeCl<sub>3</sub>
(3) FeSO<sub>4</sub>

 $(4) Mg(HCO_3)_2$ 

 $17. \qquad 3C_2H_2 \Longrightarrow C_6H_6(\ell)$ 

given that

$$G_{m}^{o}(C_{2}H_{2}) = 2.4 \times 10^{5} J$$

 $G_m^o(C_6H_6) = -1.4 \times 10^5 J$ 

Find log<sub>10</sub> k at 25°C

- **18.** 1.86 gm of aniline is converted into acetanilide with 90% efficiency. Mass of acetanilide formed is  $[X] \times 10^{-2}$  gm
- **19.** Freezing point of  $C_6H_6(\ell)$  is 5.5°C. If 10g of  $C_4H_{10}$  is mixed with 200g of  $C_6H_6(\ell)$ . Calculate freezing point of solution  $k_f = 5.12$ °C/m.
- **20.** De-broglie's wavelength of a  $\cdot$  proton and an  $\alpha$  -particle is same. Caculate ratio of their velocities
- **21.** If  $[H^+]$  changed from 1M to  $10^{-4}$  M

Find change in electrode potential  $E^{o}_{MnO_{4}^{-}/Mn^{+2}}, \left(\frac{RT}{F} = 0.059\right)$ 

 $[Assume [MnO_4^{-}] = [Mn^{+2}] = 1M]$ 

- 22. V ml of a hydrocarbon  $C_xH_y$  requires 6V ml of oxygen for complete combustion & forms 4V ml of  $CO_2$ . Determine y
- **23.** Sucrose  $\xrightarrow{\text{lorder}}$  Glucose + Fructose

t<sub>1/2</sub> = 3.33 hour

f = fraction remaining of sucrose at 9 hour.

Find out value of  $100 \times \log\left(\frac{1}{f}\right)$ 

 $[\log_{10} 2 = 0.3]$